

Brief Communications

Synthesis of phosphinic and phosphonic analogs of homoserine

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A convenient procedure for the synthesis of 1-amino-3-hydroxypropylphosphinic and -phosphonic acids (analogs of homoserine) was developed. The procedure involves the reaction of salts of phosphinic and phosphonic analogs of *S*-methylmethionine with AcONa/AcOH followed by hydrolysis.

Key words: analogs of homoserine, 1-amino-3-hydroxypropylphosphinic and -phosphonic acids.

Phosphinic analogs of amino acids and their metabolites exhibiting biological activities have been extensively studied in recent years. Thus α -amino- γ -methylthiopropylphosphinic acid, which is an analog of methionine, possesses high antibacterial activity.¹ However, a phosphinic analog of another γ -substituted α -amino acid, *viz.*, of homoserine, which plays a major role in metabolism of methionine and *S*-adenosylmethionine (the main donor of methyl groups), has not hitherto been known.

The present study was aimed at developing a convenient procedure for the synthesis of 1-amino-3-hydroxypropylphosphinic and -phosphonic acids (**1a** and **1b**, respectively), which are analogs of homoserine.

α -Aminoalkylphosphinic acids are generally prepared by the addition of H_3PO_2 ,² $\text{HP}(\text{SiOMe})_2$,³ or $(\text{EtO})_2\text{CHP}(\text{O})(\text{H})\text{OEt}$ **4** at the C=N bond of *N*-substituted imines followed by deprotection as well as by reactions of oximes with H_3PO_2 .⁵ Previously, a phosphonic analog of homoserine **1b** has been prepared by a multistage procedure⁶ involving C-alkylation of *N*-pro-

tected ethyl aminomethylphosphonate, which is an analog of glycine, with *O*-protected iodoethanol followed by deprotection.

However, according to our data, these procedures appeared to be of little use in the synthesis of amino acid **1a**. Thus the reaction of 3-hydroxypropanal oxime with H_3PO_2 afforded the target amino acid in a yield of only 1%.

Taking into account that salts of *S*-methylmethionine are readily converted into homoserine⁷ and the fact that organophosphorus analogs of *S*-methylmethionine **3a** and **3b** are readily accessible,⁸ we examined the possibility of the use of the latter for the synthesis of compounds **1a** and **1b**, respectively. Initially, the phosphinic and phosphonic analogs of methionine, *viz.*, **2a** and **2b**, were converted into sulfonium salts **3a** and **3b**, respectively. Compound **3a** was prepared according to a procedure reported previously.⁸ It appeared that phosphonic acid **3b** was also more conveniently prepared by methylation of sulfide **2b** under the action of methyl *p*-toluenesulfonate according to a procedure used for the synthesis

We found that heating of sulfonium salts **3a** and **3b** with NaOAc in AcOH followed by treatment with aqueous HCl gave rise to amino(hydroxy)alkylphosphinic and -phosphonic acids **1a** and **1b**, respectively, in 80–85% yields. Apparently, the reactions proceeded through intermediate formation of the corresponding *O*-acetyl derivatives.

B. Bromine (0.15 mL) was added to a stirred solution of phosphinic acid **1a** (278 mg, 2 mmol) in a mixture of concentrated HBr (1 mL) and EtOH (5 mL). Then the reaction mixture was stirred at 20 °C for 30 min, propylene oxide was

added, and the mixture was kept at +4 °C for 2 h. The precipitate that formed was filtered off and washed with EtOH. After recrystallization from aqueous EtOH and drying *in vacuo* over P₂O₅/KOH, phosphonic acid **1b**, which was identical with the authentic sample prepared according to a procedure A, was obtained in a yield of 260 mg (84%).

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References

1. J. G. Dingwall, in *Proc. III Int. Conf. Chem. and Biotech. Biol. Active Comps.*, Bulgaria, Sofia, 1985, 1, 87.
2. E. K. Baylis, C. D. Campbell, and I. G. Dingwall, *J. Chem. Soc., Perkin Trans. 1*, 1984, 1, 2845.
3. D. Grobenly, *Synthesis*, 1987, 942.
4. L. Maier and P. J. Diel, *Phosphorus, Sulfur, and Silicon*, 1991, **57**, 57.
5. R. M. Khomutov and T. I. Osipova, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1978, 1954 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1978, **27**, 1722 (Engl. Transl.)].
6. R. Jacquier, F. Ouazzani, M.-L. Roumestant, and Ph. Viallefont, *Phosphorus and Sulfur*, 1988, **36**, 73.
7. T. F. Lavine, N. F. Floyd, and M. S. Cammaroti, *J. Biol. Chem.*, 1954, **207**, 107.
8. T. I. Osipova, A. R. Khomutov, Yu. N. Zhukov, and R. M. Khomutov, *Izv. Akad. Nauk, Ser. Khim.*, 1999, 1360 [*Russ. Chem. Bull.*, 1999, **48**, 1348 (Engl. Transl.)].
9. W. B. Lawson, E. Gross, C. M. Foltz, and B. Witkop, *J. Am. Chem. Soc.*, 1961, **83**, 1509.
10. K. Kahr and C. Berther, *Chem. Ber.*, 1960, **93**, 132.

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